The information contained harein is lumished without warranty of any kind. Employees should use this information only as a survitement to other information orthour due to orthout without warranty of any kind. Employees should use this information only as a survitement to other information orthour due to orthout without warranty of any kind.		A	Comparison of the second secon	
Amproved hy and the Effective Date Second Forlabouritory use only. Nal for day, too				1.0.T. Sodium hudrov
Other Precautions And Structure and Andrews When working with chemicals.	DEC 281995	D		
SECTION IX SPECIAL PRECAUTIONS Precautions to be Taken in Handling & Storing Keep container tightly closed when not in use. Store in corrosion-proof area. Isolate from incompatible materials. Wash thoroughly after handling.	n form an explosive	jen gas, which ca	Jnusual Fire and Explosion Hazards Peacts with most metals to produce hydrogen gas, which can form an explosive mixture with air.	Jnusual Fire and Explosion Hazards
Lab coat, protective clothing.				
Ventilation Locett Exmand X special Ventilation Mechanical (denoral) other other Protective. Gloves Rubber gloves Eye Protection Safety goggles/faceshield				
Protection NIOSH approved respirator if TL	an a	his material.	Special Firefighting Procedures	Special Firefighting Pr Flood with water, d
	ounding fire.	propriate for surr	Use extinguishing media appropriate for surrounding fire	
Waste Disposal Method Dispose in accordance with all applicable federal, state and local environmental regulation	N/A N/A	Flammable Limits in Air % by Volume		Flash Point (Method Used)
dilute HCI. Flush area with flooding amounts of water. (Use caution.)	DATA	s, odorless liquid. AND EXPLOSION HAZARD DATA	Colorless, odorless liquid FIRE AND EXPLOS	Appearance & Odor
Steps to be Taken in Case Material is Released or Spilled			Complete	Solubility in Water
IQ			N/A	Vapor Density (Air=1)
	=1) N/A	Evaporation Rate	(g) N/A	Vapor Pressure (mm Hg)
Hazardous Polymerization Conditions to Avoid	>95%	Percent Volatile by Volume (%)	N/A	Boiling Point ("F)
Hazardous N/A	N/A	Specific Gravity (H ₂ O=1)	N/A	Melting Point ("F)
(Materials to Avoid) Strong acids, organic materials, most common metals, zinc, aluminum, magnesium, halogenated hydrocarbons.			* chemical subject to the reporting requirements of SARA Title III. SECTION III	* chemical subject to the repo
Section VI REACTIVITY DATA Stability Conditions to Avoid Stable Lunstable N/A				
	OF MIXTURES PEL. TLV Units 2 mg/m ³ 2 mg/m ³		HAZARDOUS INGREDIENTS >>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>>	Section II
Call a physician. If swallowed , do not induce vomiting; if conscious, give large amounts of water. Follow with diluted vinegar, fruit juice or whites of eggs, beaten with water. If	See Section II		SOLUTION	
Emergency and First Aid Procedures	NaOH in H ₂ O	Formula N	SODIUM HYDROXIDE, 0.1N	roduct Name SOL
eyes. Liquid may cause permanent eye damage. Ingestion: Ingestion may cause severe burning to mouth and stomach. Ingestion may cause nausea and vomiting.	NAME Caustic Soda Solution	SECTION I Chemical Synonyms	3 CHEMICAL	EAD CAREFULLY BEFORE USING CHEMICAL SHA requires that this form be kept on file. 'roduct No. SA 9746 M
Inhalation: Inhalation of vapors may cause severe irritation or burns of the respiratory system, pulmonary edema, or lung inflammation. Liquid may cause burns to skin and			RAT Ety data sheet	MATERIAL SAFETY
	Health Hazard	1111日日日日日日日日日日日日日日日日日日日日日日日日日日日日日日日日日	·	letal.